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## Structure Reports

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Liang-Zhong Xu, ${ }^{\mathbf{a} *}$ Kai Li, ${ }^{\text {a }}$ Hai-Bin Song, ${ }^{\text {b }}$ Yong-Wei Huang $^{\mathrm{a}}$ and Kai Zhou ${ }^{\text {a }}$
${ }^{\text {a College of Chemistry and Molecular }}$ Engineering, Qingdao University of Science and Technology, Qingdao 266042, People's Republic of China, and ${ }^{\mathbf{b}}$ State Key Laboratory and Institute of Elemento-Organic, Chemistry, Nankai University, Tianjin 300071, People's Republic of China

Correspondence e-mail: qknhs@163169.net

## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.005 \AA$
$R$ factor $=0.051$
$w R$ factor $=0.132$
Data-to-parameter ratio $=13.5$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## (Z)-3-Phenyl-2-(propan-2-ylidenehydrazono)-thiazolidin-4-one

In the title compound, $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{OS}$, the molecular backbone formed by the thiazolidinone and 2-(propan-2-ylidene)hydrazone groups is planar. There are some weak intra- and intermolecular hydrogen-bond interactions which stabilize the crystal structure.

## Comment

It has been reported that both thiazoles and thiazolidines have good and wide insecticidal, fungicidal, herbicidal and acaricidal activity (Melnikov, et al., 1979; Kratt \& Salberk, 1984). As part of our search for compounds with good herbicidal and fungicidal activity, the title compound, (I), was synthesized. We report here its crystal structure.

(I)

Bond lengths and angles in the thiazolidine ring (Table 1) are in agreement those reported previously (Buyuktimkin et al., 1984; Yang et al., 1995; Pomés Hernandez et al., 1996). The molecular backbone formed by the thiazolidinone and 2 -(propan-2-ylidene)hydrazone groups is essentially planar, the maximum deviations being -0.070 (3) and $0.029(3)^{\circ}$ for C 2 and N 2 , respectively. The phenyl ring is twisted by $70.98(7)^{\circ}$ with respect to this plane. There are some weak $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{N}$ intra- and intermolecular hydrogen-bond interactions which stabilize the crystal structure (Table 2).

## Experimental

A mixture of hydrazine ( 0.02 mol ), ethyl chloroacetate ( 0.02 mol ) and phenyl isothiocyanate ( 0.02 mol ) was stirred in refluxing acetone $(5 \mathrm{ml})$ for 5 h at 327 K to afford the title compound ( 3.71 g , yield $75 \%$ ). Single crystals suitable for X-ray measurements were obtained by recrystallization from ethanol at room temperature.

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## Crystal data

$\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{OS}$
$M_{r}=247.31$
Monoclinic, $P 2_{2} / c$
$a=16.859(9) \AA$
$b=5.667(3) \AA$
$c=12.580(7) \AA$
$\beta=90.685(9)^{\circ}$
$V=1201.8(11) \AA^{3}$
$Z=4$
$D_{x}=1.367 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 1528
reflections
$\theta=2.4-22.4^{\circ}$
$\mu=0.26 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Block, yellow
$0.18 \times 0.16 \times 0.12 \mathrm{~mm}$
Data collection
Bruker SMART CCD area-detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
$T_{\text {min }}=0.950, T_{\text {max }}=0.970$
6139 measured reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.051$
$w R\left(F^{2}\right)=0.132$
$S=0.89$
2121 reflections
157 parameters
H -atom parameters constrained


Figure 1
View of (I), with displacement ellipsoids drawn at the $40 \%$ probability level.

All H atoms were placed in calculated positions, with $\mathrm{C}-\mathrm{H}=0.93-$ $0.97 \AA$, and refined using a riding model, with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$ for the aryl and methylene H atoms and $1.5 U_{\mathrm{eq}}(\mathrm{C})$ for the methyl H atoms.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

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